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Tarik Al-Shemmeri - Funmaza Display

1st law of thermodynamics $1 \text{ kJ} = 0.94782 \text{ btu}$ $1 \text{ btu} = 1.055056 \text{ kJ}$ The Conservation of Energy Principle Rate of energy, work or heat transfer: The amount of energy gained by a system is equal to

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1.4 Independent and Dependent Thermodynamic Properties

The state postulate indicates that an intensive property of a pure substance can be determined from two independent

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Introduction Applied Thermodynamics - II Classification of Thermodynamic Cycles Application of Thermodynamics Important areas are Power generation and Refrigeration. Usually accomplished by systems that operate on a thermodynamic cycle.

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